



INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

<p>(51) International Patent Classification⁴ : C03C 23/00, C03B 19/00, 15/00</p>	<p>A1</p>	<p>(11) International Publication Number: WO 86/ 02919 (43) International Publication Date: 22 May 1986 (22.05.86)</p>
<p>(21) International Application Number: PCT/GB85/00499 (22) International Filing Date: 4 November 1985 (04.11.85) (31) Priority Application Number: 8427915 (32) Priority Date: 5 November 1984 (05.11.84) (33) Priority Country: GB (71) Applicant (for all designated States except US): TSL THERMAL SYNDICATE P.L.C. [GB/GB]; P.O. Box 6, Neptune Road, Wallsend, Tyne and Wear NE28 6DG (GB). (72) Inventor; and (75) Inventor/Applicant (for US only) : WINTERBURN, John, Alexander [GB/GB]; 7 Ellesmere Gardens, Tynemouth, North Shields NE30 3BE (GB). (74) Agent: NEWBY, John, R.; J.Y. & G.W. Johnson, Furnival House, 14-18 High Holborn, London WC1V 6DE (GB).</p>		<p>(81) Designated States: AT (European patent), BE (European patent), BG, BR, CH (European patent), DE (European patent), FI, FR (European patent), GB (European patent), HU, IT (European patent), JP, KR, LU (European patent), NL (European patent), RO, SE (European patent), US. Published With international search report.</p>
<p>(54) Title: VITREOUS SILICA</p> <p>(57) Abstract</p> <p>An improved quality vitreous silica body and/or improved quality product made at high temperature in a vitreous silica vessel is/are obtained by applying a polarising potential across the boundary surfaces of the vitreous silica body or vessel to cause migration of impurity ions away from one of the boundary surfaces thereof. Single crystal silicon (10) of reduced alkali content is drawn from melt (12) in a vitreous silica crucible (14) with a polarising voltage applied across the wall of the crucible.</p> <div data-bbox="927 1287 1373 1814"> </div>		

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	GA	Gabon	MR	Mauritania
AU	Australia	GB	United Kingdom	MW	Malawi
BB	Barbados	HU	Hungary	NL	Netherlands
BE	Belgium	IT	Italy	NO	Norway
BG	Bulgaria	JP	Japan	RO	Romania
BR	Brazil	KP	Democratic People's Republic of Korea	SD	Sudan
CF	Central African Republic	KR	Republic of Korea	SE	Sweden
CG	Congo	LI	Liechtenstein	SN	Senegal
CH	Switzerland	LK	Sri Lanka	SU	Soviet Union
CM	Cameroon	LU	Luxembourg	TD	Chad
DE	Germany, Federal Republic of	MC	Monaco	TG	Togo
DK	Denmark	MG	Madagascar	US	United States of America
FI	Finland	ML	Mali		
FR	France				

Vitreous silica.

This invention relates to a method of improving the quality of a product made from or made in contact with a body of vitreous silica.

In one aspect, the invention relates to a method
5 of making an improved vitreous silica product, such as a crucible, a tube or a plate, having a reduced impurity content, and to a product produced by the method.

The invention also relates to an improved method
10 for using a vitreous silica vessel for the high temperature processing of a material contained therein. Particularly valuable uses of this method of the invention are found in the treatment of molten semiconductor materials in a vitreous silica crucible (e.g. in the drawing of a single crystal of silicon).

Discussion of Prior Art

15 An arc moulded crucible (AMC) is normally produced by fusing a powder material under the influence of an electrical arc while the powder material (e.g. quartz powder) is held in place in a rotating mould (e.g. of water cooled metal) by centrifugal force, with or without
20 the application of a vacuum via the mould wall(s).

A vitreous silica crucible is commonly used to contain a melt from which a single crystal is drawn. In the case of semiconductor materials, high purity is of vital importance and much effort has been, and is still being, given to avoiding impurity contamination of the
25 melt during the crystal pulling operation. The crucible is one potential source of such contamination.

In the case of the pulling of a silicon crystal

- 2 -

from melt in a vitreous silica crucible, alkali impurities may transfer to the molten silicon from the wall(s) of the crucible and a variety of different, often time-consuming and otherwise expensive, procedures have been
5 proposed for the purification of the starting material used for a vitreous silica crucible to reduce the alkali impurities therein.

It has now been discovered that a reduced impurity content in material fused in a vitreous silica crucible
10 can be obtained by the simple expedient of applying a polarising potential across the wall(s) of the crucible, at least while the latter is at a temperature in excess of 700°C, for such a period and with such a polarity that ions of the or each impurity will migrate across
15 the wall(s) of the crucible away from the inside surface thereof.

Impurity migration can be effected when the crucible is first fused from the starting material and/or by applying a correctly polarised potential across the wall(s)
20 of the crucible when it contains melt. In the case of alkali impurity ions (which so far appear to be the impurities most easily removed from the crucible wall/melt interface) a voltage with positive polarity on the inside and negative polarity on the outside of the crucible
25 will be required. Voltages between 1 and 2000 volts appear to be effective over the temperatures and times typical for pulling a single crystal from a bath of molten semiconductor material. The migration rate is a function of temperature and applied voltage and preferably the
30 temperature of the inside surface of the crucible wall(s) is in excess of 900°C.

The crucible wall(s) is/are not a significant barrier to the diffusion of impurities (e.g. Na, K or Li) which in the absence of an electrolysing potential may migrate

- 3 -

through the wall(s) from the crucible holder, and which in the presence of an electrolysing potential can migrate into the wall(s) from melt in the crucible.

Although the invention is thought to have an important commercial impact in the areas of the high temperature production of products from vitreous silica vessels and the manufacture of such vessels with reduced alkali impurity content, it will be appreciated that vitreous silica tubing whose impurity content has been reduced by electrolysis can have other useful applications than as an intermediary in the manufacture of a vitreous silica vessel.

Hence it should be appreciated that the invention also extends to the production of improved vitreous silica products (e.g. tubing) either by a one-stage process in which an ion-migrating potential is applied across the product during manufacture, or by a two-stage process in which an ion-migrating potential is applied across a heated billet prior to further processing (e.g. drawing) with or without further electrolysis during that further processing.

Once ion migration in a vitreous silica body has been achieved using the principles described herein, it would often be desirable to remove any ion-enhanced region from the body so that even if back diffusion should subsequently occur, there will be a net improvement in impurity content of the vitreous silica body.

Summary of the Invention

Thus, in its broadest aspect, the present invention relates to a method of improving the quality of a product made from, or made in contact with, a body of vitreous silica, which is characterised in that impurity ions are made to migrate away from one boundary surface of

the body towards an opposite boundary surface thereof by applying a polarising potential across the boundary surfaces of the body while the body is maintained at a temperature above 700°C.

- 5 The polarising potential is typically poled to make alkali metal ions migrate away from the said one boundary surface. Typical temperatures are in the range 800 to 2000°C, typical process times are from one to a few tens of hours at the lower end of the range and a few minutes
10 to an hour at higher temperatures and typical polarising potentials from a few tens of volts to a few kilovolts.

The method can be used to make tubing and other vitreous silica products of improved purity.

- It is thus feasible to electrolyse tubing during
15 fusing to cause migration of ions away from one wall surface and then to produce the desired product either from a powder starting material taken from ion-depleted regions of the tubing (e.g. by grinding off the ion-rich region(s)), or directly from the electrolysed tubing
20 using ion-depleted regions for the inside walls. Preferably a polarising potential should be reapplied every time the wall material is heated above about 750°C and certainly above 900°C if some back-diffusion is not to occur but if a sufficient impurity depletion has been
25 achieved during manufacture of the product or its starting material, such back-diffusion may not matter. In practice, however, since electrolysing a vitreous silica vessel during use is neither difficult nor expensive to achieve, it could be employed even where a low-alkali-impurity
30 content vitreous silica material was used in its manufacture.

Brief Description of the Drawings

The invention will now be more fully described with reference to the accompanying drawings, in which:

- 5 -

Figure 1 is a schematic indication of the invention applied to improve single crystal semiconductor material drawn from a vitreous silica crucible and as described in the following Examples 10 to 12,

5 Figure 2 shows a vitreous silica crucible whose purity has been improved by the method of the invention which is being processed as described in the following Examples 1 and 2,

10 Figure 3 shows how a vitreous silica crucible can be made in accordance with the invention in the manner described in the following Examples 3 and 4,

Figures 4 and 5 show a crucible being made in the manner described, respectively, in the following Examples 5 to 7 and Example 8, and

15 Figure 6 shows how the quality of vitreous silica tubing can be improved as described in the following

Description of Preferred Embodiments

Figure 1 shows, purely schematically, a single crystal 10 of silicon being drawn from a bath 12 of molten silicon contained in a vitreous silica arc moulded crucible 14. In the usual way, the crucible 14 is contained within a graphite susceptor 16 heated by an induction coil (not shown) and the crystal 10 depends from a seed crystal 10a which has been slowly drawn upwards out of the surface of the bath 12.

A d.c. potential is applied between the seed crystal 10a and the susceptor 16 with the polarity as shown, this potential being maintained throughout the pulling operation. The magnitude of the applied potential can vary from a minimum which is sufficient to overcome con-

tact resistances and ionisation potentials and establish a current of a few microamps, to a maximum where the high voltage causes arcing or other problems. In practice, a potential of between a few volts and a few hundred
5 volts would normally be used. Figure 1 is further discussed in the following Examples 10 to 13.

Examples of the Invention

Example 1: (see Figure 2)

A previously manufactured crucible 20 of 300 mm diameter with a 6 mm wall thickness was heated to 1050°C
10 in nitrogen. An internal electrode 21, consisting of high purity carbon powder loosely filled into the crucible 20, and an external electrode was provided by a graphite holder 22 into which the crucible 20 fitted loosely. The gap between the holder 22 and the crucible 20 was
15 filled with high purity carbon powder 23. Care was taken to ensure excess carbon powder did not cause short circuits. The upper 10 mm of the crucible protruded above the holder and internal powder fill to act as a barrier to surface tracking by the applied voltage. The holder
20 was made the negative electrode.

The voltage was applied gradually on the crucible reaching 1050°C to keep the electrolysing current below 50 mA. After 40 minutes, the full voltage of 2.5 KV could be applied.

25 Electrolysis continued for 4 hours, then the temperature was allowed to drop to room temperature with the full voltage still applied.

Analysis of the crucible material after this treatment is shown in Table 1 as AMC 3.

30 Example 2:

A second crucible was treated as in Example 1 except

- 7 -

that the polarising voltage was switched off at 800°C as the crucible was cooling following electrolysis at 1050°C.

Analysis of the crucible is shown in Table 1 as
5 AMC 4. From the results it can be seen that back diffusion of the alkali ions is insignificant below 800°C.

Example 3: (see Figure 3)

A further crucible 30 was heated directly in a spinning graphite mould 31 using an oxy propane flame 32 so
10 that the crucible softened and came into intimate contact with the mould. The mould was made negative and the burner positive using a voltage of 3.8 KV. The high electrical impedance of the flame greatly reduced the voltage available for electrolysis but some improvement
15 was measured as can be seen from the figures shown in Table 1 as AMC 10. The time of electrolysis was 5 minutes.

Example 4:

A further crucible was treated as in Example 3 except that an R.F. induction plasma replaced the flame 32.
20 The analysis of this treated crucible is shown as AMC 14 in Table 1.

Example 5: (see Figure 4)

A crucible 40 was manufactured using the spinning mould method to hold high purity quartz powder in position.
25 Heating was with an arc 41 and the mould 42 was of water-cooled metal. An electrolysing voltage of 10 KV was applied for the final 2 minutes of heating using the arc 41 as the positive electrode and the mould 42 as the negative electrode. The analytical result is shown
30 as AMC 21 in Table 1.

- 8 -

Example 6:

A crucible was manufactured as for Example 5 except that an uncooled graphite mould was used in place of the water-cooled mould 42. The analytical result is shown in Table 1 as AMC 23. The improved result when compared with Example 5 is believed to be due to the higher electrical resistance of the quartz powder kept cold by the water-cooled mould in Example 5, reducing the voltage available for electrolysis.

10 Example 7:

A crucible was manufactured as for Example 6 except that during fusion a partial vacuum of 8 - 7 kPa was applied between the mould 42 and the forming crucible 40 via a pipe 45.

15 The analytical result is shown as AMC 31. The lower impurity content in this case is believed to be due to partial ionisation of the gas in the gap between the mould and the forming crucible due to the partial vacuum and this ionised gas acting as the negative electrode.

TABLE 1

Crucible	Position	Surface	Noted impurity in ppm by weight		
			Na ₂ O	K ₂ O	Li ₂ O
AMC 3	Side Wall	Inner	<0.1	<0.1	<0.1
		Outer	1.5	4.2	3.3
	Base Wall	Inner	0.2	0.2	0.2
		Outer	1.6	3.3	2.8
AMC 4	Side Wall	Inner	<0.1	<0.1	<0.1
		Outer	2.5	2.0	4.4
	Base Wall	Inner	<0.1	<0.1	<0.1
		Outer	0.1	0.2	0.3
AMC 10	Mean Value	Inner	0.9	1.5	0.2
		Outer	2.2	2.0	2.0
AMC 14	Mean Value	Inner	<0.1	<0.1	<0.1
		Outer	0.2	0.4	0.2
AMC 21	Mean Value	Inner	0.9	1.2	0.2
		Outer	1.4	2.5	2.0
AMC 23	Mean Value	Inner	0.2	1.3	<0.1
		Outer	0.9	2.5	0.5
AMC 31	Mean Value	Inner	<0.1	0.2	<0.1
		Outer	0.1	0.4	0.1
AMC 33	Mean Value	Inner	<0.1	<0.1	<0.1
		Outer	0.2	0.5	<0.1
Starting Material for AMC 3, 4, 10, 14			5.6	5.4	3.8
Starting Material for AMC 21, 23, 31			1.4	2.5	3.2

- 10 -

Example 8: (see Figure 5)

A manufactured crucible 50 was placed over a closely fitting graphite internal mould 51 and heated externally with an oxy-propane ribbon burner 52. The temperature reached on the surface of the crucible was sufficient to remelt it.

A potential difference was applied between the burner 52 and the mould 51 of 4.5 KV. The mould was rotated at 1 RPM so that the flame swept over all the crucible. The mould was the positive electrode. During the processing it was noted that the flame was coloured by the ions being electrolysed from the crucible.

An additional result of the heating was that the outside of the crucible became glazed.

The analytical results appear in Table 1 as AMC 33.

Example 9: (see Figure 6)

A cylindrical pipe 60 of fused quartz with an external diameter of 200 mm, a length of 1500 mm and a wall thickness of 25 mm was subject to electrolysis across the wall with a potential difference of 10KV.

An inner electrode 61 (anode) was made by coating the inner surface with a layer approximately 2 mm thick consisting of a paste of low alkali titanium dioxide (British Titan Products Ltd. - Grade A-HR) and a proprietary low alkali silica sol (Nalfloc Ltd. - "Nalcoag 1034A"). Connection to this electrode was made with a nickel chromium alloy band 62 which was a spring fit in the bore.

An outer electrode (cathode) consisted of a layer 63 approximately 2 mm thick coating the whole outer surface except the ends and consisting of a paste of ferric oxide and a silica sol. Connection to this electrode

- 11 -

was made with an open pitch coil 64 of nickel chromium heat resisting wire.

Electrolysis was carried out at 1050°C. On reaching temperature, the voltage was gradually increased so as
5 to avoid exceeding the current limitation of the power supply (100 ma). Maximum voltage of 10 KV was reached after 8 hours 40 minutes.

Electrolysis continued for 30 hours, when the furnace was allowed to cool naturally. The voltage was switched
10 off when the pipe had cooled to 800°C.

The results of the electrolysis are shown as B4 in Table 2.

Example 10:

The cylindrical pipe of fused quartz from Example
15 9 was machined externally to remove 1 mm from the bore and 3 mm from the external surface leaving a wall thickness of 21 mm. After cleaning with detergent and dilute hydrofluoric acid it was reheated in a graphite resistance furnace and drawn into tubing. Some of this tubing was
20 reworked with flames on a glass working lathe to the form of a crucible. The analysis of the crucible is shown in Table 2 as C1.

- 12 -

TABLE 2

Surface		Impurity concentration in ppm by weight		
		Na ₂ O	K ₂ O	Li ₂ O
B4	Inner	< 0.1	< 0.1	< 0.1
	Centre	< 0.1	< 0.1	< 0.1
	Outer	13.9	4.8	8.0
Tube from B4		0.1	0.1	< 0.1
C1	Inner	< 0.1	< 0.1	< 0.1
	Outer	< 0.1	0.1	< 0.1
Starting Material for B4		8.8	4.3	4.4

Example 11: (see Figure 1)

A single crystal "puller" was modified to allow a voltage to be applied between the silicon single crystal 10a of Figure 1 and the graphite susceptor 16 during the growing operation which was carried out in Argon at 1 atmosphere gauge. After the crystal 10 had achieved the desired diameter, a voltage of 50 - 1000 V was applied between the crystal (positive) and the susceptor (negative). The voltage was derived from a current limited source of 0.010A.

Results for the silicon crystal 10 are shown in Table 3.

Example 12:

A single crystal was grown as for Example 11 except that a sub-atmospheric pressure of 15-20 torr was used

- 13 -

in the puller. The maximum polarising voltage was limited to 200 V.

The results are shown in Table 3 (S2).

Example 13:

5 A single crystal was grown as for Example 11 except that a coating of glassy carbon (obtained by the pyrolysis of propane diluted with Argon) had been made on the outside of the quartz crucible before use in order to improve the electrical contact between the crucible and the
10 susceptor.

The results are shown in Table 3 (S3).

TABLE 3

Resistivity of Single Crystal
ohm-cm P.Type Silicon

Without polarizing Voltage		Crystal Shoulder	145
		Crystal end	115
With polarizing Voltage	S1	Crystal Shoulder	500
		Crystal end	300
	S2	Crystal Shoulder	850
		Crystal end	600
	S3	Crystal Shoulder	1250
		Crystal end	1050

- 14 -

From the foregoing Examples, it will be appreciated that the electrolysing temperature and time conditions are related one to the other and to the wall thickness across which the polarising potential is applied. In
5 summary, these process conditions are preferably that the body is maintained for a time of at least 1 hour/mm wall thickness in the temperature range 800° - 1200°C and at least 1 min/mm thickness in the temperature range 1201° - 2000°C. The effective polarising potential
10 applied across the boundary surfaces preferably exceeds 10V/mm thickness but not 1 KV/mm thickness.

- 15 -

CLAIMS

1. A method of improving the quality of a product made from, or made in contact with, a body of vitreous silica, characterised in that impurity ions are made to migrate away from one boundary surface of the body
5 towards an opposite boundary surface thereof by applying a polarising potential across the boundary surfaces of the body while the body is maintained at a temperature above 700°C.
2. A method as claimed in claim 1, characterised
10 in that the temperature of the body is above 900°C and the polarising potential is poled to make alkali metal ions migrate away from the said one boundary surface.
3. A method as claimed in claim 1 or claim 2, characterised in that the body is cooled from a tempera-
15 ture in excess of 1000°C and the polarising potential is maintained in the same polarity across the boundary surfaces until the temperature of the body is less than 800°C.
4. A method as claimed in any preceding claim,
20 characterised in that the body is maintained for a time corresponding to the thickness of the wall being polarised of at least 1 hour/mm thickness in the temperature range 800° - 1200°C and at least 1 min/mm thickness in the temperature range 1201° - 2000°C and the effective polar-
25 ising potential applied across the boundary surfaces exceeds 10V/mm thickness.
5. A method as claimed in claim 4, characterised
in that the polarising potential applied across the boundary surfaces is less than 1KV/mm thickness.
- 30 6. The employment of the method claimed in any of claims 1 to 4 to improve the purity of melt at a tem-

- 16 -

perature above 900°C and contacting the inner wall of a vitreous silica containing vessel, characterised in that the polarising potential is applied via the melt in a direction to cause impurity ions in the vitreous silica to migrate away from said inner wall.

7. The method of claim 6, characterised in that the melt is a semiconductor material and the polarising potential is applied thereto via a single crystal of the semiconductor material in contact with the melt.

8. The method of claim 6 or 7, characterised in that the vitreous silica vessel has its outer surface in electrical contact with carbon and the polarising potential is applied across the wall of the vessel via the melt and the carbon.

9. The method of any of claims 6 to 8, characterised in that the polarising potential lies in the range 1 to 2000 volts and the melt is at positive potential relative to the outer wall of the vessel.

10. The employment of the method claimed in any of claims 1 to 4, to reduce the concentration of impurity ions in a tube of vitreous silica characterised in that the polarising potential is applied in the radial direction across the wall of the tube while the latter is at a temperature above 800°C to cause impurity ion migration in the radial direction, and in that the wall thickness of the tube is subsequently reduced by removing tube material at least from that surface of the tube towards which the ions migrated.

11. The method of claim 10, characterised in that a polarising potential in the range 500 to 10000 volts is applied across the wall of the tube between electrically conductive coatings applied to said walls, and in

- 17 -

that vitreous silica material is removed from both walls when the electrolysis of the tube has been completed.

12. The employment of the method of any of claims 1 to 4, to produce an improved vitreous silica crucible, characterised in that the crucible is produced by a method which involves building up the thickness of a layer of fused vitreous silica against an electrically conducting mould member which defines one surface of the crucible, and in that said mould member is used as one electrode by which the polarising voltage is applied across the layer to cause impurity ion migration in one of the directions towards and away from the mould member.

13. The method of claim 12, characterised in that the other electrode for applying the polarising potential across the layer of fused vitreous silica is a hot gas or arc used to fuse the layer of vitreous silica.

14. The method of claim 12, characterised in that the hot gas is generated by a combustion gas burner.

15. The method of claim 12, characterised in that the hot gas is generated by a plasma torch or electric arc.

16. The method of any preceding claim, characterised in that the ion migration occurs with the vitreous silica in an inert gas or reducing gas environment.

17. The method of any preceding claim, characterised in that the ion migration occurs with the vitreous silica at a pressure different from atmospheric pressure.

18. The employment of the method of claim 1 to purify an arc moulded crucible of vitreous silica material characterised in that one wall of the crucible is raised to a temperature in excess of 800°C with a hot gas and a polarising potential is applied to the crucible to cause impurity ion migration from one wall into the gas.

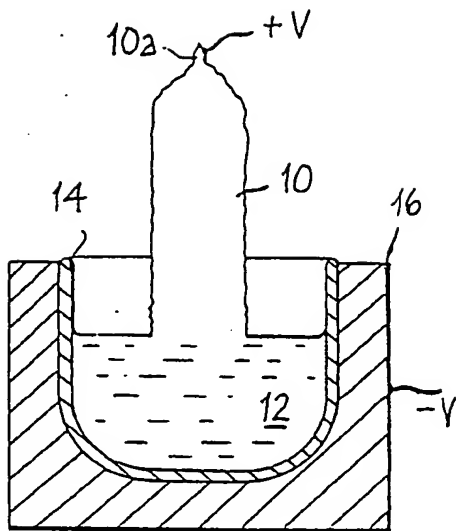


FIG. 1

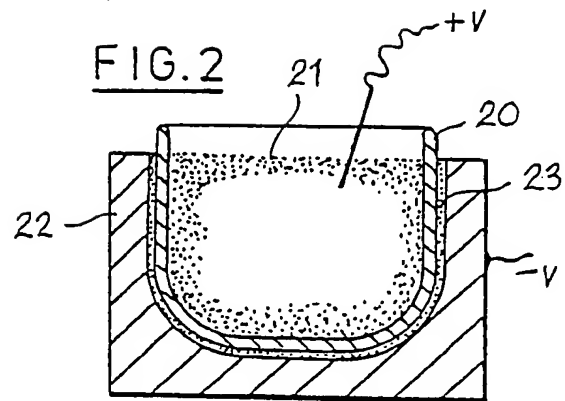


FIG. 2

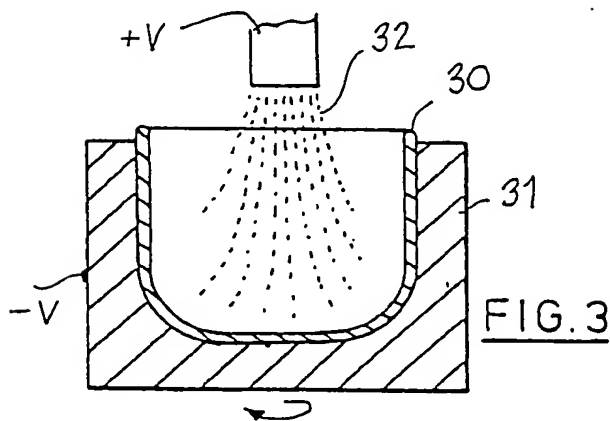


FIG. 3

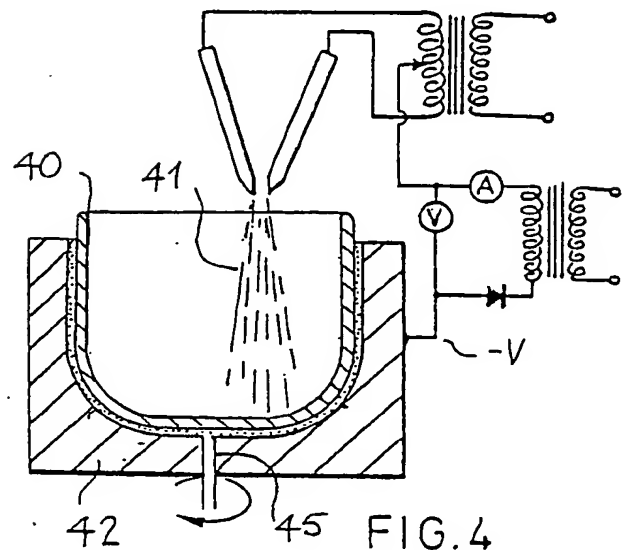


FIG. 4

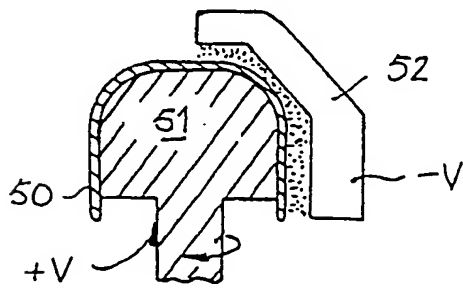


FIG. 5

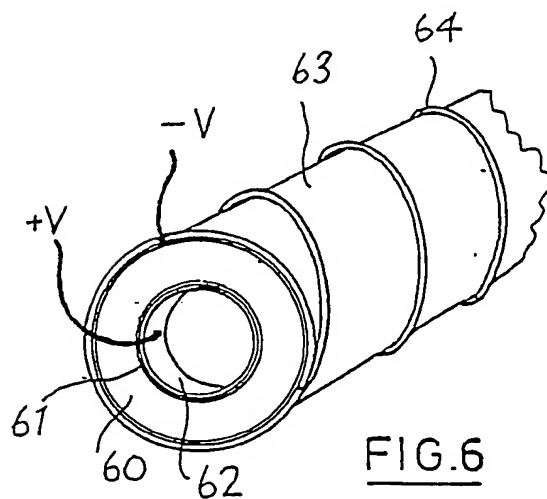





FIG. 6

INTERNATIONAL SEARCH REPORT

International Application No PCT/GB 85/00499

I. CLASSIFICATION OF SUBJECT MATTER (if several classification symbols apply, indicate all) * According to International Patent Classification (IPC) or to both National Classification and IPC IPC ⁴ : C 03 C 23/00; C 03 B 19/00; C 30 B 15/00																				
II. FIELDS SEARCHED <div style="text-align: right; font-size: small;">Minimum Documentation Searched ⁷</div> <table border="1" style="width: 100%; border-collapse: collapse;"> <thead> <tr> <th style="width: 25%; text-align: left; font-size: small;">Classification System</th> <th style="text-align: left; font-size: small;">Classification Symbols</th> </tr> </thead> <tbody> <tr> <td style="vertical-align: top;">IPC⁴</td> <td>C 03 C 23/00 C 03 B 19/00 C 30 B 15/00</td> </tr> </tbody> </table> <div style="text-align: center; font-size: x-small; margin-top: 5px;">Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched ⁸</div>			Classification System	Classification Symbols	IPC ⁴	C 03 C 23/00 C 03 B 19/00 C 30 B 15/00														
Classification System	Classification Symbols																			
IPC ⁴	C 03 C 23/00 C 03 B 19/00 C 30 B 15/00																			
III. DOCUMENTS CONSIDERED TO BE RELEVANT ⁹ <table border="1" style="width: 100%; border-collapse: collapse;"> <thead> <tr> <th style="width: 10%; font-size: x-small;">Category ⁹</th> <th style="width: 70%; font-size: x-small;">Citation of Document, ¹¹ with indication, where appropriate, of the relevant passages ¹²</th> <th style="width: 20%; font-size: x-small;">Relevant to Claim No. ¹³</th> </tr> </thead> <tbody> <tr> <td style="text-align: center; vertical-align: top;">X, Y</td> <td>FR, A, 1363233 (CORNING GLASS WORKS) 4 May 1964, see abstract; figure 1; page 2, right-hand column, paragraph 3 --</td> <td style="text-align: center; vertical-align: top;">1-9, 12-15</td> </tr> <tr> <td style="text-align: center; vertical-align: top;">X</td> <td>FR, A, 1123264 (PROSILIS) 19 September 1956, see abstract; figure 2; page 1, right-hand column --</td> <td style="text-align: center; vertical-align: top;">1-5, 10-11</td> </tr> <tr> <td style="text-align: center; vertical-align: top;">X</td> <td>DE, B, 1047390 (J. LIETZ) 24 December 1958, see the whole document --</td> <td style="text-align: center; vertical-align: top;">1, 16</td> </tr> <tr> <td style="text-align: center; vertical-align: top;">Y</td> <td>FR, A, 2480272 (HERAEUS QUARZSCHMELZE) 16 October 1981, see claims; figures --</td> <td style="text-align: center; vertical-align: top;">1, 6-9</td> </tr> <tr> <td style="text-align: center; vertical-align: top;">A</td> <td>US, A, 1910017 (P.L. HULIN) 23 May 1933 -----</td> <td></td> </tr> </tbody> </table>			Category ⁹	Citation of Document, ¹¹ with indication, where appropriate, of the relevant passages ¹²	Relevant to Claim No. ¹³	X, Y	FR, A, 1363233 (CORNING GLASS WORKS) 4 May 1964, see abstract; figure 1; page 2, right-hand column, paragraph 3 --	1-9, 12-15	X	FR, A, 1123264 (PROSILIS) 19 September 1956, see abstract; figure 2; page 1, right-hand column --	1-5, 10-11	X	DE, B, 1047390 (J. LIETZ) 24 December 1958, see the whole document --	1, 16	Y	FR, A, 2480272 (HERAEUS QUARZSCHMELZE) 16 October 1981, see claims; figures --	1, 6-9	A	US, A, 1910017 (P.L. HULIN) 23 May 1933 -----	
Category ⁹	Citation of Document, ¹¹ with indication, where appropriate, of the relevant passages ¹²	Relevant to Claim No. ¹³																		
X, Y	FR, A, 1363233 (CORNING GLASS WORKS) 4 May 1964, see abstract; figure 1; page 2, right-hand column, paragraph 3 --	1-9, 12-15																		
X	FR, A, 1123264 (PROSILIS) 19 September 1956, see abstract; figure 2; page 1, right-hand column --	1-5, 10-11																		
X	DE, B, 1047390 (J. LIETZ) 24 December 1958, see the whole document --	1, 16																		
Y	FR, A, 2480272 (HERAEUS QUARZSCHMELZE) 16 October 1981, see claims; figures --	1, 6-9																		
A	US, A, 1910017 (P.L. HULIN) 23 May 1933 -----																			
<div style="font-size: x-small;"> <p>* Special categories of cited documents: ¹⁰</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> </div>		<div style="font-size: x-small;"> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"&" document member of the same patent family</p> </div>																		
IV. CERTIFICATION <table border="1" style="width: 100%; border-collapse: collapse;"> <tr> <td style="width: 50%; padding: 5px;"> Date of the Actual Completion of the International Search <div style="text-align: center;">5th February 1986</div> </td> <td style="width: 50%; padding: 5px;"> Date of Mailing of this International Search Report <div style="text-align: center;">21 FEB. 1986</div> </td> </tr> <tr> <td style="width: 50%; padding: 5px;"> International Searching Authority <div style="text-align: center;">EUROPEAN PATENT OFFICE</div> </td> <td style="width: 50%; padding: 5px;"> Signature of Authorized Officer <div style="text-align: center;">  M. VAN MOL </div> </td> </tr> </table>			Date of the Actual Completion of the International Search <div style="text-align: center;">5th February 1986</div>	Date of Mailing of this International Search Report <div style="text-align: center;">21 FEB. 1986</div>	International Searching Authority <div style="text-align: center;">EUROPEAN PATENT OFFICE</div>	Signature of Authorized Officer <div style="text-align: center;">  M. VAN MOL </div>														
Date of the Actual Completion of the International Search <div style="text-align: center;">5th February 1986</div>	Date of Mailing of this International Search Report <div style="text-align: center;">21 FEB. 1986</div>																			
International Searching Authority <div style="text-align: center;">EUROPEAN PATENT OFFICE</div>	Signature of Authorized Officer <div style="text-align: center;">  M. VAN MOL </div>																			

ANNEX TO THE INTERNATIONAL SEARCH REPORT ON

INTERNATIONAL APPLICATION NO.

PCT/GB 85/00499 (SA 11187)

This Annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report. The members are as contained in the European Patent Office EDP file on 13/02/86

The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
FR-A- 1363233		None	
FR-A- 1123264		CH-A- 338566 BE-A- 545590 GB-A- 783244 US-A- 2897126	
DE-B- 1047390		None	
FR-A- 2480272	16/10/81	DE-A,C 3014311 JP-A- 56149333 GB-A,B 2075967 US-A- 4416680 CH-B- 651533	22/10/81 19/11/81 25/11/81 22/11/83 30/09/85
US-A- 1910017		None	

For more details about this annex :
see Official Journal of the European Patent Office, No. 12/82